

20장 열역학 제 1법칙

$$11. W = -P \left( \frac{nRT_h}{P} - \frac{nRT_c}{P} \right) = -nR(T_h - T_c)$$

$$W = -nR\Delta T = -(0.200 \text{ mol})(8.314 \text{ J/mol} \cdot \text{K})(280 \text{ K}) = -466 \text{ J}$$

$$12. P = \left( \frac{P_i}{V_i} \right) V,$$

$$(a) W = - \int_i^f P dV = - \int_{V_i}^{3V_i} \left( \frac{P_i}{V_i} \right) V dV$$

$$W = - \left( \frac{P_i}{V_i} \right) \left[ \frac{V^2}{2} \right]_{V_i}^{3V_i} = - \frac{P_i}{2V_i} (9V_i^2 - V_i^2) = -4P_i V_i$$

$$(b) PV = nRT,$$

$$\left[ \left( \frac{P_i}{V_i} \right) V \right] V = nRT$$

$$\therefore T = \left( \frac{P_i}{nRV_i} \right) V^2$$

13.

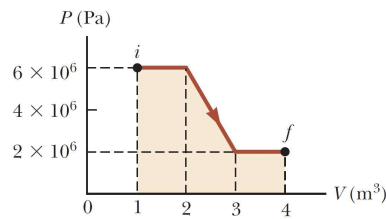


그림 P20.13

$$(a) W = - \int P dV$$

$$= -(6.00 \times 10^6 \text{ Pa})(2.00 \text{ m}^3 - 1.00 \text{ m}^3) - (4.00 \times 10^6 \text{ Pa})(3.00 \text{ m}^3 - 2.00 \text{ m}^3) \\ - (2.00 \times 10^6 \text{ Pa})(4.00 \text{ m}^3 - 3.00 \text{ m}^3)$$

$$\therefore W_{i \rightarrow f} = -12.0 \text{ MJ}$$

$$(b) W_{f \rightarrow i} = +12.0 \text{ MJ}$$

18.

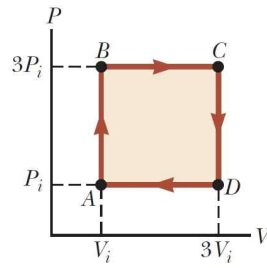


그림 P20.18

(a)  $W = W_{AB} + W_{BC} + W_{CD} + W_{DA}$   
 $= 0 - 3P_i(3V_i - V_i) + 0 - P_i(V_i - 3V_i) + 0$   
 $= -4P_iV_i = -4nRT_i$   
 $\therefore W = -4(1.00 \text{ mol})(8.314 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K})(273 \text{ K}) = 9.08 \text{ kJ}$

(b) 초기온도와 마지막 온도는 동일  
 $\therefore \Delta E_{\text{int}} = 0$ , 그리고  $Q = -W = 9.80 \text{ kJ}$