

21장 기체의 운동론

1. (a) $V = \frac{4}{3}\pi r^3 = \frac{4}{3}\pi(0.150 \text{ m})^3 = 1.41 \times 10^{-2} \text{ m}^3$

$PV = nRT$ 에서,

$$n = \frac{PV}{RT} = \frac{(1.013 \times 10^5 \text{ N/m}^2)(1.41 \times 10^{-2} \text{ m}^3)}{(8.314 \text{ N} \cdot \text{m/mol} \cdot \text{K})(293 \text{ K})} = 0.588 \text{ mol}$$

분자의 수는 $N = nN_A$ 이다.

$$\begin{aligned} N &= nN_A = (0.588 \text{ mol})(6.02 \times 10^{23} \text{ molecules/mol}) \\ &= 3.54 \times 10^{23} \text{ helium atoms} \end{aligned}$$

(b) $\bar{K} = \frac{1}{2}m_o\bar{v}^2 = \frac{3}{2}k_bT$

$$= \frac{3}{2}(1.38 \times 10^{-23} \text{ J/K})(293 \text{ K}) = 6.07 \times 10^{-21} \text{ J}$$

(c) He 원자의 질량 $m_o = \frac{M}{N_A} = \frac{4.0026 \text{ g/mol}}{6.02 \times 10^{23} \text{ molecules/mol}}$

$$= 6.65 \times 10^{-24} \text{ g} = 6.65 \times 10^{-27} \text{ kg}$$

$$v_{rms} = \sqrt{\bar{v}^2} = \sqrt{\frac{2\bar{K}}{m_o}} ; = \sqrt{\frac{2 \times 6.07 \times 10^{-21} \text{ J}}{6.65 \times 10^{-27} \text{ kg}}} = 1.35 \text{ km/s}$$

4. 압력과 분자 운동 에너지와의 관계식 (21.15 참고) : $P = \frac{2N}{3V} \left(\frac{m_o\bar{v}^2}{2} \right)$

$$\bar{K} = \frac{1}{2}m_o\bar{v}^2 = \frac{3PV}{2N} \text{ 에서 } N = nN_A$$

$$\begin{aligned} \bar{K} &= \frac{3PV}{2nN_A} = \frac{3(8.00 \text{ atm})(1.013 \times 10^5 \text{ Pa/atm})(5.00 \times 10^{-3} \text{ m}^3)}{2(2 \text{ mol})(6.02 \times 10^{23} \text{ molecules/mol})} \\ &= 5.05 \times 10^{-21} \text{ J} \end{aligned}$$

10. (a) $Q = nC_p\Delta T$ 이용, $\Delta T = 420\text{ K} - 300\text{ K} = 120\text{ K}$

$$Q = nC_p\Delta T = (1.00\text{ mol})(28.8\text{ J/mol} \cdot \text{K})(420\text{ K} - 300\text{ K}) = 3.46\text{ kJ}$$

(b) $\Delta E_{\text{int}} = nC_V\Delta T = (1.00\text{ mol})(20.4\text{ J/mol} \cdot \text{K})(120\text{ K}) = 2.45\text{ kJ}$

(c) $\Delta E_{\text{int}} = Q + W$ 에서

$$W = -Q + \Delta E_{\text{int}} = -3.46\text{ kJ} + 2.45\text{ kJ} = -1.01\text{ kJ}$$

14. (a) 이상기체의 단열과정에서 P 와 V 의 관계에서,

$$P_i V_i^\gamma = P_f V_f^\gamma \text{ 이용}$$

$$P_f = P_i \left(\frac{V_i}{V_f} \right)^\gamma = (5.00\text{ atm}) \left(\frac{12.0}{30.0} \right)^{1.40} = 1.39\text{ atm}$$

(b) $T_i = \frac{P_i V_i}{nR} = \frac{5.00(1.013 \times 10^5\text{ Pa})(12.0 \times 10^{-3}\text{ m}^3)}{(2.00\text{ mol})(8.314\text{ J/mol} \cdot \text{K})} = 366\text{ K}$

$$T_f = \frac{P_f V_f}{nR} = \frac{1.39(1.013 \times 10^5\text{ Pa})(30.0 \times 10^{-3}\text{ m}^3)}{(2.00\text{ mol})(8.314\text{ J/mol} \cdot \text{K})} = 253\text{ K}$$

(c) 단열과정에서 $Q = 0$

(d) $\Delta E_{\text{int}} = nC_V\Delta T$ 에서, $C_V = \frac{R}{\gamma-1} = \frac{2}{5}R$

$$\begin{aligned} \Delta E_{\text{int}} &= nC_V\Delta T = (2.00\text{ mol}) \left[\frac{2}{5}(8.314\text{ J/mol} \cdot \text{K}) \right] (253\text{ K} - 366\text{ K}) \\ &= -4.66\text{ kJ} \end{aligned}$$

(e) $W = \Delta E_{\text{int}} - Q = -4.66\text{ kJ} - 0 = -4.66\text{ kJ}$